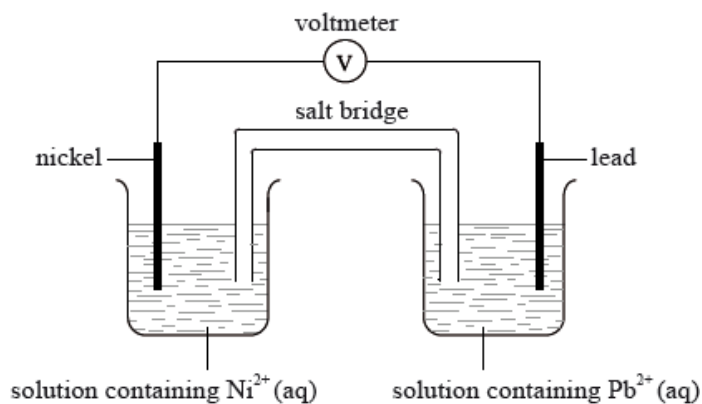
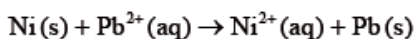

SL Paper 1

The overall reaction in the voltaic cell below is:



Which statement is correct for the nickel half-cell?

- A. Nickel is the positive electrode (cathode) and is reduced.
 - B. Nickel is the negative electrode (anode) and is reduced.
 - C. Nickel is the positive electrode (cathode) and is oxidized.
 - D. Nickel is the negative electrode (anode) and is oxidized.
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