

Markscheme

May 2018

Chemistry

On-screen examination



11 pages

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The following are the annotations available to use when marking responses.

Annotation	Explanation	Shortcut	Annotation	Explanation	Shortcut
~	Correct point, place at the point in the response where it is clear that the candidate deserves the mark	Alt+1	NBOD	No benefit of the doubt	Alt+4
AEr	Arithmetic error		NEX	No explanation given	
BOD	Benefit of the doubt	Alt+3	NGE	Not good enough	
λ	Omission, incomplete	Alt+7	0	Not worthy of any marks	
CON	Contradiction	Alt+6	NWS	No working shown	
	Valid part (to be used when more than one element is required to gain the mark)		T	Test box used for additional marking comments	
ECF	Error carried forward	Alt+8	?	Unclear	Alt+2
0	Dynamic annotation, it can be expanded to surround work		SEEN	Seen; must be stamped on all blank response areas	Alt+9
~~~	Horizontal wavy line that can be expanded		~~~	Vertical wavy line that can be expanded	
	Highlight tool that can be expanded to mark an area of a response		WITE	Words to that effect	
NAQ	Not answered the question		✓ 1 ✓ 2 ✓ 3 ✓ 4	Award 1, 2, 3, 4 marks. For use in holistically marked questions only	

## **Markscheme instructions**

- 1 Mark positively. Give candidates credit for what they have achieved and what is correct. Do not deduct marks for incorrect responses.
- 2 Follow the markscheme provided and award only whole marks.
- 3 Each marking point appears on a separate line.
- 4 The maximum mark for each subpart is indicated in the "Total" column.
- 5 Where a mark is awarded a tick should be placed in the text at the precise point where it is clear the candidate deserves the mark.
- 6 Each marking point in a question part should be awarded separately unless there is an instruction to the contrary in the Notes column.
- 7 A question subpart may have more marking points than the total allows. This will be indicated by the word "**max**" in the Answer column. Further guidance may be given in the Notes column.
- 8 Additional instructions on how to interpret the markscheme are in **bold** italic text in the Answer column.
- 9 Alternative wording may be indicated in the Answer column by a slash (/). Either alternative is equally acceptable but the candidate cannot be rewarded for both as they are associated with the same marking point.
- 10 Alternative answers are indicated in the Answer column by "*or*". Either alternative is equally acceptable but the candidate cannot be rewarded for both as they are associated with the same marking point.
- 11 If two related points are required to award a mark, this is indicated by "*and*" in the answer column.
- 12 Words in brackets () in the Answer column are not necessary to gain the mark.
- **13** Words that are <u>underlined</u> are essential for the mark.
- 14 In some questions a reverse argument is also acceptable. This is indicated by the abbreviation *ORA* (or reverse argument) in the Notes column. Candidates should not be rewarded for reverse arguments unless *ORA* is given in the Notes column.
- 15 If the candidate's response has the same meaning or is clearly equivalent to the expected answer the mark should be awarded. In some questions this is emphasized by the abbreviation *WTTE* (words to that effect) in the Notes column.
- 16 When incorrect answers are used correctly in subsequent question parts the follow through rule applies. Award the mark and add *ECF* (error carried forward) to the candidate response.
- 17 The order of marking points does not have to be the same as in the Answer column unless stated otherwise.
- 18 Marks should not be awarded where there is a contradiction in an answer. Add CON to the candidate response at the point where the contradiction is made.
- 19 Do not penalize candidates for errors in units or significant figures unless there is specific guidance in the Notes column.
- 20 Questions with higher mark allocations will generally be assessed using a level response method using task specific clarifications developed with reference to the criteria level descriptors. A candidate's work should be reviewed to determine holistically the mark for each row of the holistic grid and a mark awarded for each row.

Question	Answers	Notes	Total	Criterion
1 a	Group     Period       Carbon:     4     2       Hydrogen:     1     1       Oxygen:     6     2	Award only <b>1 mark</b> if groups and periods are interchanged Award <b>1 mark</b> only if all groups are correct Award <b>1 mark</b> only if all periods are correct	3	A
b	methanoic (acid)		1	А
C	Correct bonds shown <i>or</i> all lone pairs correct structure fully correct	Accept dots or crosses or any combination	2	A
d	$HCOOH(aq) + H_2O(l) \rightleftharpoons H_3O^+(aq) + HCOO^-(aq)$ reactant(s) correctproducts correctstates all correctcorrect use of equilibrium arrow	Allow $HCOOH(aq) \rightleftharpoons H^{+}(aq) + HCOO^{-}(aq)$ Accept incorrect order for example, $H_2CO_2, CHOOH$ Accept I for HCOOH Allow any double-headed arrow	4	A
е	add/react with a base/alkali		1	A

2	а	protons = 12			
				2	A
	_	neutrons = 12			-
	b	Mg ₃ (PO ₄ ) ₂	Brackets must be present	1	A
	С	for any relevant property, for example: state, metal or non-metal, malleable etc property 1 identified	Do <b>not</b> accept reference to valence electrons, solubility or type of element		
		statement of how Mg <b>and</b> Cl <b>or</b> Cl ₂ are distinguished	Accept linked answers in switched response boxes	4	А
		property 2 identified			
		statement of how Mg and CI or Cl2 are distinguished			
	d	ionic bond	Accept electrovalent	1	A
	е	sulphate ion identified	Accept sulphuric acid		
		calcium sulphate is insoluble	Accept Ca ions	4	A
		magnesium sulphate is soluble <b>or</b> Mg ²⁺ (ions) remain in solution	Accept Mg ions		
		filter (calcium sulphate from the solution) or remove precipitate or solid			
	f	2.5 (moles)		1	А
3	а	CaCl ₂		1	Α
	b	Independent variable: <u>Percentage</u> of calcium carbonate or type of bone/tooth Dependent variable:	Do <i>not</i> accept calcium		
		<ul> <li>(Volume or amount of) carbon dioxide produced</li> <li>Accept any two reasonable control variables, for example</li> <li>mass of bone/tooth</li> <li>volume or amount of acid used</li> <li>concentration of acid used</li> <li>temperature</li> <li>surface area or size of bone</li> </ul>	Do <b>not</b> accept amount of bone Do <b>not</b> accept type of acid	4	В

⊟ Table Object						
Animal Part	Volume of carbon dioxide produced (ml) when reacting to h	hydrochloric acid				
20.0	Trial 1	Trial 2	Trial 3			
Elephant tooth						
Elephant bone						
Horse tooth						
Horse bone						
Human tooth	•					
Human bone						
Monkey tooth						
Monkey bone						
Dolphin tooth					Λ	C
Dolphin Bone					-	0
indication of mo column for inde column for depe	pre than one trial pendent variable: type of bone endent variable: volume of CO2			ECF from part (b)		

			1	2	3		4		
		Additional equipment (E)Eq butMethod (M)Att but relationMethod (M)Pla of tool	Equipment suggested but is not relevant	Equipment to measure mass <i>or</i> volume <i>or</i> one control variable	Equipment to measure mass <i>and</i> volume <i>and</i> one control variable Method is described, could be followed by another student producing relevant data Plans to use at least three samples of each type of tooth/bone		ure Equipment to measure <b>nd</b> mass <b>and</b> volume <b>and</b> two control variables		
4 ;			Attempt at a method but may not be relevant	Attempt at a method but detail is insufficient for another student to follow and is not likely to give relevant data			Complete method is described, fully explained and could be repeated by another student		
			Plans to test one type of tooth/bone	Plans to test each type of tooth/bone			Plans to use at least three samples of each type of tooth/bone <b>and</b> calculates a mean	17	В
		Assumptions (A)	Assumptions that all bone/teeth/CaCO ₃ has reacted <b>or</b> temperature <b>or</b> pressure remain constant	Assumptions that all bone/teeth/CaCO ₃ has reacted <b>and</b> temperature <b>or</b> pressure remain constant	Assumptions t bone/teeth/Ca reacted <b>and</b> temperature <b>a</b> pressure rema constant	hat all CO₃ has <i>nd</i> ain			
		Safety (S)	Use of relevant safety equipment	Use of relevant safety equipment linked to corrosive hazard from acid					
4	а	these points are out	iers/anomalies <b>or</b> reason v	why an outlier was obtained		eg incorre amount of	ect amount of CaCO ₃ was used ^c CaCO ₃ was too high <b>and</b> too low	2	С
		repeat				Do <b>not</b> ac	cept interpolation		
	b	use of graph to give	mass of 2.40 g						
		evidence of calculati	on of % by mass			seen or in	nplied		
		correct value of 24.7	,			ECF mark	king point 2	4	С
		correct identification	of horse bone			ECF mark	king point 3 – do <b>not</b> award this e	3	

5	а	Image 1: Corrosive  Image 2: Flammable Image 3: Toxic	Award <b>1 mark</b> for each	3	В
	b	Image 1		1	В
	С	appropriate average = 56 (cm ³ ) it is not appropriate to include an outlier in the average	Award <b>1 mark only</b> for inappropriate average of 60. Do <b>not</b> award 2 nd marking point even if justification is correct.	2	С
	d	Limestone = $1.33 \pm 0.03$ (cm ³ s ⁻¹ ) Crushed ovster shell = $0.28 \pm 0.03$ (cm ³ s ⁻¹ )		2	С
	е	limestone particles are smaller than crushed oyster shell	ORA		С
		(so) rate of reaction is greater for limestone	ORA	3	
		correct use of terms surface area and rate of reaction			D
	f	the crushed oyster shell takes longer to be broken down			
		allowing more time for the hen to absorb <b>or</b> use the calcium carbonate for eggshell production	Accept "digest"	2	С
	g	limestone data is less reliable			
		(because) greater variability in data between trials <b>or</b> poor consistency			0
		crushed oyster shell <u>data</u> is more reliable		4	C
		(because) data have good agreement <b>or</b> low variability <b>or</b> good consistency			
	h	increase number of trials or		1	С
		use same size particles of limestone and oyster shell			Ũ

6	а	A						1	A
	b	more than two comp boiling point (depend molecules with small	ounds <b>or</b> mixtures of com ls on size of molecules) <b>o</b> est mass <b>or</b> length will ha	pounds can be separated <b>r</b> change in state from liquid ve the lowest boiling point	I to gas	WTTE		3	D
	С		1	 I			1		
			1	2	3		4		
		Method (M)	Comment about removal	Comment about more than one method of removal	Describes rem all three metho	noval using ods	Describes removal using all three methods linked to science Skimmer – oil and water are immiscible Burning – fumes, smoke Dispersion – components remain in water, affect aquatic life	o	
		Advantages/disa dvantages (AD)	One advantage <b>or</b> disadvantage implied	One advantage <b>and</b> one disadvantage for one method implied <b>or</b> One advantage <b>or</b> one disadvantage for more than one method implied	Advantages <b>a</b> disadvantages for all three m	<i>nd</i> s identified ethods		o	U
		Appraisal (A)	"Best method identified" <b>and</b> supported by scientific evaluation						

	1	2	3	4	
Pipeline (distribution of oil) (P)	A comment about a pipeline	One advantage <b>or</b> one disadvantage of a pipeline implied	One advantage <b>and</b> one disadvantage of a pipeline implied	More than one advantage <b>and</b> more than one disadvantage of a pipeline	
Alternative methods (AM)	A comment about an alternative method	One advantage <b>or</b> one disadvantage of one alternative method	One advantage <i>or</i> one disadvantage of both alternative methods <i>or</i> More than one advantage <i>and</i> more than one disadvantage of one alternative method	More than one advantage <b>and</b> more than one disadvantage of a both alternative methods	14
Environmental considerations (E)	An environmental impact	More than one environmental impact	More than one environmental impact with at least one supported by science		
Social considerations (S)	A social impact of transporting oil	More than one social impact of transporting oil			
Appraisal (A)	A concluding appraisal linking the issues discussed				