

Biology

Standard level

Paper 1B

Markscheme

Question	Answers	Notes	Marks
1. a	<ul style="list-style-type: none"> - Growth rates decrease in both species when grown together (Species A: 2.5 → 1.2 cm/week; Species B: 3.0 → 1.8 cm/week). ✓ - Seed production decreases (Species A: 200 → 80; Species B: 250 → 120). ✓ 	Award 1 mark per correct comparison. Numerical data must be included for full marks.	2
1. b	<ul style="list-style-type: none"> - Demonstrates interspecific competition ✓ - Both species compete for resources, reducing growth/reproduction. ✓ <p>OR</p> <ul style="list-style-type: none"> - Competitive exclusion principle ✓ - One species may dominate over time. ✓ 	Accept either explanation. Award marks for linking data to the principle.	2
1. c (i)	<ul style="list-style-type: none"> - Abiotic factor (e.g., low pH) favors one species ✓ 	Award 1 for general influence, 1 for specific example.	1
1. c (ii)	<ul style="list-style-type: none"> - Fundamental niche: Full range of conditions a species can occupy without competition. ✓ - Realized niche: Restricted due to competition (e.g., Species A avoids areas with B). ✓ 	Award 1 for definition, 1 for effect, 1 for example.	2

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2. a (i)	<ul style="list-style-type: none"> - Inverse relationship: As water temperature increases, oxygen solubility decreases. ✓ OR - Negative correlation shown on the graph. ✓ 	<p>Award 1 mark for a clear description of the trend.</p> <p>Do not accept "they are related" without specifying the trend.</p>	1
2. a (ii)	<ul style="list-style-type: none"> - Rising temperatures reduce oxygen solubility, limiting availability for aquatic organisms. ✓ - This can lead to hypoxia, stressing respiration/metabolism (e.g., fish gills less efficient). ✓ 		2
2. b	<ul style="list-style-type: none"> - Polar water molecules form weak interactions (dipole-induced dipole) with O₂. ✓ - Hydrogen bonding creates "cages" that trap nonpolar O₂, enhancing solubility. ✓ 		2
2. c	<ul style="list-style-type: none"> - CO₂ is more soluble than O₂ because it reacts with water to form carbonic acid (H₂CO₃). ✓ - Biological significance: CO₂ solubility affects pH (ocean acidification), impacting coral calcification/shellfish survival. ✓ 		2

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3. a	<ul style="list-style-type: none"> - Energy flows linearly and is lost as heat (second law of thermodynamics). ✓ - Sun is the constant energy source; heat cannot be recycled. ✓ - Matter (e.g., carbon) is recycled via biogeochemical cycles (e.g., decomposition). ✓ 	<p>Accept any two.</p> <p>Do not accept "energy is reused" without qualification.</p>	2
3. b (i)	- 10% (10,000 → 1,000 kJ/m ² /yr). ✓		1
3. b (ii)	- 10% (100 → 10 kJ/m ² /yr). ✓		1
3. c	<ul style="list-style-type: none"> - CO₂ release: Decomposers respire, returning carbon to the atmosphere. ✓ - Nutrient recycling: Break organic matter into inorganic forms (e.g., CO₂, minerals) for producers. ✓ 	<p>Accept "humus formation" or "detritivore actions" if linked to carbon.</p>	2
3. d	<ul style="list-style-type: none"> - Increased photosynthesis: Higher CO₂ boosts primary production. ✓ - Trophic cascade: More energy flows to consumers (↑ biomass). ✓ - Climate impact: Excess CO₂ may disrupt ecosystems (e.g., coral bleaching). ✓ 	Accept any two.	2

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4. a	- Complementary base pairing: A-T and C-G pairing ensures accurate template copying; mismatches are rare due to hydrogen bonding specificity. ✓		1
4. b	- DnaB helicase unwinds the double helix, separating parental strands to serve as templates. ✓ - This exposes bases for complementary pairing, enabling synthesis of new strands (semi-conservative mechanism). ✓	Accept "creates replication fork" or "allows polymerase access."	2