

$$1) a) c = Td\sqrt{g} \quad = 5 \text{ m m}^{\frac{1}{2}} \text{ s}^{-1} \\ = \text{m}^{\frac{3}{2}}$$

b) Absolute uncertainty in measurement of 20 oscillations is same as for one. This is divided by 20 so % uncertainty is less.

c) i) [Straight line touching as many points as possible]

ii) [Should now attempt to extrapolate line to vertical axis]

Since  $T = \frac{c}{d\sqrt{g}}$ ,  $T \propto \frac{1}{d}$  so the data should show a proportional relation. This is consistent as line is straight and through the origin.

d) [Find gradient using a large triangle]

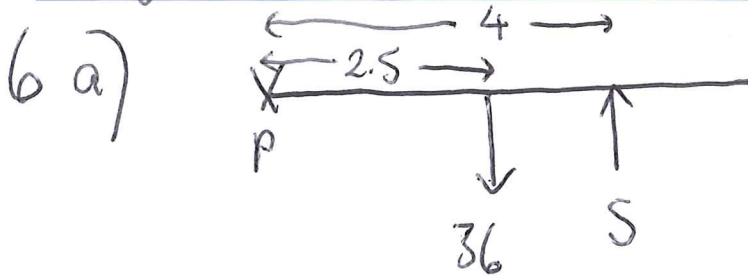
$$\text{Gradient } 0.51 \leq m \leq 0.57$$

$$\text{Gradient} = \frac{c}{\sqrt{g}}$$

$$g = \left( \frac{c}{\text{gradient}} \right)^2$$

$$8.6 \leq g \leq 10.7 \text{ ms}^{-2}$$

- 2.a) Rearranging:  $m = \frac{VI t}{L_v}$  If  $V$  and  $I$  are constant then  $m$  is proportional to  $t$ . Constant power provided ( $\checkmark$ )
- b) Heat will be lost so energy provided will be larger than expected ( $VI t$ ). This gives a larger value of  $L_v$ .
- c) Heat loss is a systematic error for each experiment. Finding the difference will cancel this systematic error.



$$\begin{aligned} \curvearrowleft P &= P \curvearrowright \\ 48 &= 2.5 \times 36 \\ S &= 22.5 \text{ N} \end{aligned}$$

b) i.  $\Gamma = I \alpha$

$$36 \times 2.5 = 30.6 \alpha$$

$$\alpha = 2.94 \text{ rad s}^{-2}$$

ii. The angular acceleration is not constant as the component of the weight acting perpendicular to the rod is changing and so torque is not constant.

c) i. [Knowing that motion equations will not work!]

Conservation of energy:  $\Delta GPE = \Delta KE_{\text{rot}}$

$$mg \rightarrow WL = \frac{1}{2} I \omega^2$$

vertical height drop  
by centre of mass

$$\omega = \sqrt{\frac{WL}{I}} = \sqrt{\frac{36 \times 5}{30.6}}$$

$$= 2.435 \text{ rad s}^{-1}$$

ii.  $L = I \omega = 30.6 \times 2.43$   
 $= 74.4 \text{ kg m}^2 \text{ s}^{-1}$

worry show that

7.a) i.  $C \rightarrow A$  is adiabatic

$$P_C V_C^{5/3} = P_A V_A^{5/3}$$

$$P_C = 2.8 \times 10^6 \times \left( \frac{1 \times 10^{-4}}{1.85 \times 10^{-4}} \right)^{5/3}$$

$$= 1.00 \times 10^6 \text{ Pa}$$

ii.  $A \rightarrow B$  is isothermal so  $T_B = T_A$

$$\frac{T_C}{V_C} = \frac{T_B}{V_B}$$

$$T_C = \frac{T_B}{V_B} V_C = \frac{385 \times 1.85 \times 10^{-4}}{2.8 \times 10^{-4}} = 254.4 \text{ K}$$

b) Thermal energy transferred = work done +  $\Delta U$

first law  $\rightarrow$

$$p \Delta V = 1.00 \times 10^6 (2.8 \times 10^{-4} - 1.85 \times 10^{-4})$$

$$= 95 \text{ J} \quad [\text{Negative since } W \text{ on gas}]$$

$$\Delta U = \frac{3}{2} p \Delta V = \frac{3}{2} \times 95 = 142.5 \quad [\text{Negative since cooling}]$$

$$Q = \Delta U + p \Delta V$$

$$= -142.5 - 95 = -238 \text{ J}$$

c) i. efficiency =  $\frac{288-238}{288}$  ← net work done [enclosed area]  
= 0.17

ii.  $S = \frac{\Delta Q}{T}$  ← decreases from B → C  
[constant C → A since adiabatic and increases from A → B]