

c.2 Cellular Respiration

CELLULAR RESPIRATION EQUATION (Aerobic)

$C_6H_{12}O_6 + 6O_2 \rightarrow 6CO_2 + 6H_2O + ENERGY (ATP & HEAT)$

OXIDATION A general type of chemical reaction resulting in products with lower potential energy than the reactants

REDUCTION

A general type of chemical reaction resulting in products with higher notential enerou than the reactants

REDOX

Cellular Respiration is considered to be a **REDOX** reaction: when **OXIDATION** & **REDUCTION** occur together.

ANAEROBIC RESPIRATION

Requires glucose

Does NOT require oxygen

Occurs in cytoplasm

Mechanism: Glucose is split into two molecules of pyruvate. With inadequate oxygen pyruvate will be turned into lactic acid (in animals), and alcohol $+ CO_2$ (in yeast)

Net gain of 2 ATP (Little ATP)

Final products: Animals: ATP & Lactate II Yeast: alcohol & CO₂ **Aerobic Respiration**

Requires glucose

Requires oxygen

Occurs in cytoplasm & mitochondria

Mechanism: Glucose is split into two molecules of pyruvate. Pyruvate enters LINK REACTION. Then follows KREBS CYCLE. Then occurs Electron transport chain (ECT) in cristae of mitochondria.

Net gain of 30-34 ATP (Lots of ATP)

Final products: ATP, CO₂, & Water



TEMPERATURE

Cellular respiration utilises enzymes to catalyse various reactions, they are sensitive to changes in temperature



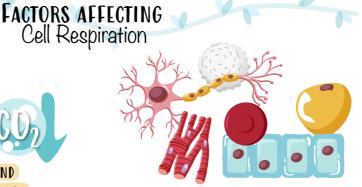
Higher concentrations of glucose increase the rate of cellular respiration



CARBON DIOXIDE

High concentrations of 0_2 will increase the rate of cellular respiration while low $C0_2$ concentrations will increase the rate

A RESPIROMETER is often used to calculate the rate of cell respiration



CELL TYPES

Different types of cells will require different amounts of energy – for example muscle cells and neurons will have higher requirements thus higher respiration rates



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