

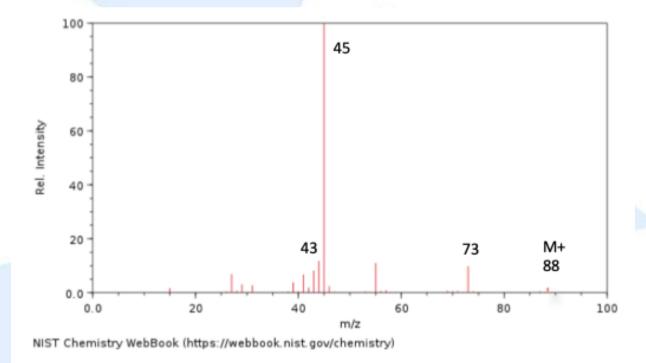
# **HL Spectroscopic identification of organic compounds: Question 17**

Identify Compound  $\mathbf{Q}$  from the data (a – d) given and explain how you arrived at your answer

### (a) Elemental analysis of compound Q

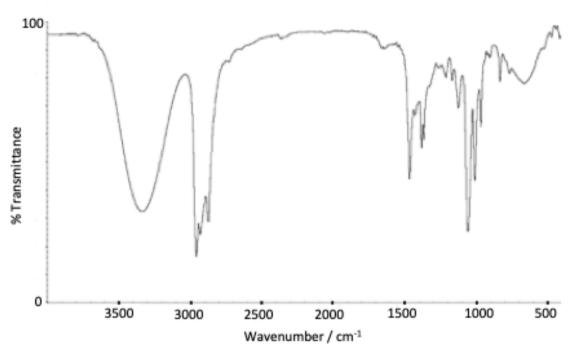
Element	Percentage by mass	
Carbon	68.11	
Hydrogen	13.74	
Oxygen	18.15	

### (b) Mass spectrum of Compound Q



#### (c) Infrared spectrum of Compound Q





Source: https://pdfs.semanticscholar.org/cf7a/7e22001c5415cf6436e95f5f11a3e3b68488.pdf

## (d) Data from <sup>1</sup>H NMR spectrum of Compound Q

Position of signal / ppm	Splitting pattern	Integration trace
0.9	doublet	6
1.1	doublet	3
1.6	complex	1
1.8	singlet	1
3.6	complex	1

<sup>&</sup>lt;sup>1</sup>H NMR data from https://spectrabase.com/spectrum/LnpPkfzUaE2